

# ETERNAL CAREER CLASSES

## SUBJECT : CHEMISTRY (Solid State)

1. Which is covalent solid ?  
(a)  $\text{SiO}_2$       (b) Diamond  
(c) Graphite      (d) all of these
2. Silicon dioxide is an example of :  
(a) metallic crystal  
(b) ionic crystal  
(c) covalent crystal  
(d) none of these
3.  $\text{ZnS}$  is  
(a) ionic crystal      (b) covalent crystal  
(c) metallic crystal      (d) van der Waals' crystal
4. Which is (are) amorphous solids ?  
(a) Rubber      (b) plastic  
(c) Glass      (d) all of these
5. How many kinds of space lattices are possible in a crystal ?  
(a) 23      (b) 7  
(c) 230      (d) 14
6. Which of the following will show anisotropy ?
7. The interparticle forces in solid hydrogen are :  
(a) hydrogen bonds  
(b) covalent bonds  
(c) co-ordinate bonds  
(d) van der Waals' forces
8. The number of octahedral in a cubical close pack array of N spheres is :  
(a)  $N / 2$       (b)  $2N$   
(c)  $4N$       (d)  $N$
9. How many tetrahedral holes are occupied in diamond ?  
(a) 25%      (b) 50%  
(c) 75%      (d) 100%
10. In which of the following crystals alternate tetrahedral voids are occupied ?  
(a)  $\text{NaCl}$       (b)  $\text{ZnS}$   
(c)  $\text{CaF}_2$       (d)  $\text{Na}_2\text{O}$

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